

# Safety Data Sheet

# **Section 01 - Product And Company Identification**

Product Identifier Hydrochloric Acid 10-35%

Other Means of Identification Aqueous hydrogen chloride, muriatic acid, hydrogen chloride, HCl, chlorohydric acid.

**Product Use and Restrictions on** 

Use

Acidizing (activation) of petroleum wells, scale removal, ore reduction, metal cleaning, pH

adjustment, industrial acidizing, generation of chlorine dioxide, regeneration of ion

exchange resins.

Initial Supplier Identifier ClearTech Industries Inc.

1500 Quebec Avenue Saskatoon, SK. Canada

S7K 1V7

Prepared By ClearTech Industries Inc. Technical Writer

Phone: 1 (800) 387-7503

**24-Hour Emergency Phone** Phone: 1 (306) 664 – 2522

## Section 02 - Hazard Identification

#### **GHS-Classification**

Skin Corrosion/IrritationCategory 1BSerious Eye Damage/IrritationCategory 1

STOT-Single Exposure Category 3

**Physical Hazards** 

Corrosive to Metals Category 1

#### **Danger**

#### **Hazards Statements**

H290 - May be corrosive to metals

H314 – Causes severe skin burns and eye damage

H335 – May cause respiratory irritation

#### **Pictograms**



#### **Precautionary Statements**

P234 - Keep only in original container

P260 - Do not breathe mist, vapours or spray.

P264 - Wash hands thoroughly after handling.

P301 +P330 + P331 - IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P303 + P361 + P353 - IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin.

P271 - Use only outdoors or in a well-ventilated area

P280 - Wear protective gloves, protective clothing, eye protection, and face protection

P390 - Absorb spillage to prevent material damage

P363 - Wash contaminated clothing before reuse

P305 + P351 + P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P304 + P340 - IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.

P310 – Immediately call a POISON CENTER or doctor/physician.

P405 - Store locked up

P403 + P233 - Store in a well-ventilated place. Keep container tightly closed

P501 – Dispose of contents/container in accordance with all federal, provincial, and/or local regulations including the Canadian Environmental Protection Act.

## Section 03 - Composition / Information on Ingredients

<b>Chemical Name</b>	<b>CAS Number</b>	Weight %	Unique Identifiers
Hydrochloric Acid	7647-01-0	10-35%	
Water	7732-18-5	65-90%	

#### Section 04 - First Aid Measures

Inhalation Remove victim to fresh air. Only give artificial respiration if breathing has stopped. If

breathing is difficult, give oxygen. Seek medical attention.

Remove contaminated clothing. Wash affected area with lukewarm water for at least 30 Skin Contact / Absorption

> minutes. If irritation persists, repeat flushing. Seek immediate medical attention. Double bag, seal, label and leave contaminated clothing, shoes and leather goods at the scene

for safe disposal.

**Eye Contact** Immediately flush the contaminated eye(s) with lukewarm, gently flowing water for at least

30 minutes, while holding the eyelid(s) open. If a contact lens is present, remove only if easy to do so. Neutral saline solution may be used as soon as it is available. Seek

immediate medical attention.

Ingestion NEVER give anything by mouth if victim is rapidly losing consciousness, is unconscious

or convulsing. Have victim rinse mouth thoroughly with water. DO NOT INDUCE

VOMITING. If vomiting occurs naturally, have victim rinse mouth with water again. Seek

immediate medical attention.

**Additional Information** This chemical is very toxic. Take proper precautions to ensure your own safety before

assisting others. DO NOT allow victim to move about unnecessarily. Symptoms of

pulmonary edema can be delayed up to 48 hours after exposure.

NOTE: Any skin or eye contact will also involve significant inhalation exposure.

# Section 05 - Fire Fighting Measures

Extinguish fire using agent suitable for surrounding fire. Cool all affected containers with Suitable Extinguishing Media

flooding quantities of water. Apply water from as far a distance as possible. Use water

spray to knock-down vapours.

**Unsuitable Extinguishing Media** Not Available

Chemical

Specific Hazards Arising From the Contact with common metals produces extremely flammable hydrogen gas. When heated or in a fire, toxic and corrosive hydrogen chloride gas is released. Hydrogen

chloride is thermally stable up to approximately 1500°C (2732°F). Above this

temperature, hydrogen chloride begins to dissociate into extremely flammable hydrogen

gas and very toxic and corrosive chlorine gas.

**Precautions for Fire-Fighters** 

**Special Protective Equipment and** Wear NIOSH-approved self-contained breathing apparatus and protective gear.

Not Available **Further Information** 

#### Section 06 - Accidental Release Measures

**Equipment / Emergency** 

**Procedures** 

Personal Precautions / Protective Wear appropriate personal protective equipment. Ventilate area. Only enter area with PPE. Stop or reduce leak if safe to do so.

**Environmental Precautions** Prevent product from entering sewers and waterways.

**Methods and Materials for** Containment and Cleaning Up SMALL SPILLS: Contain and soak up spill with absorbent material which does not react with spilled chemical. Put material in suitable, covered, labeled containers. Flush area

with water. Do not get water inside containers.

Contaminated absorbent material may pose the same hazards as the spilled product. LARGE SPILLS: Contact fire and emergency services and supplier for advice.

## Section 07 - Handling and Storage

This material is VERY TOXIC and CORROSIVE. Use proper equipment for lifting and **Precautions for Safe Handling** 

transporting all containers. Use sensible industrial hygiene and housekeeping practices. Wash thoroughly after handling. Avoid all situations that could lead to harmful exposure.

Store in a cool, dry, well-ventilated area, out of direct sunlight and away from heat sources. **Conditions for Safe Storage** 

Keep quantity stored as small as possible. Drums should be vented when received and then

at least weekly to relieve internal pressure.

Incompatibilities Metals, sodium, bases, formaldehyde, oxidizing agent, reducing agents, perchloric acid,

sulfuric acid. Potassium permanganate, aldehydes, epoxides, fluorine, acetylides, bories,

carbides, phosphide, silicides, hexalithium disilicide.

# Section 08 - Exposure Controls and Personal Protection

Component	Regulation	Type of Listing	Value
Hydrochloric Acid	ACGIH	TLV-C	2 ppm
	OSHA	PEL-T-C	5 ppm (7 mg/m <sup>3</sup> )

#### **Engineering Control(s)**

Mechanical ventilation (dilution or local exhaust), process or personnel enclosure and **Ventilation Requirements** 

control of process conditions must be provided in accordance with all fire codes and regulatory requirements. Supply sufficient replacement air to make up for air removed by

exhaust systems.

Emergency shower and eyewash must be available and tested in accordance with Other

regulations and be in close proximity.

#### **Protective Equipment**

Chemical goggles, full-face shield, or a full-face respirator is to be worn at all times when Eyes/Face

product is handled. Contact lenses should not be worn; they may contribute to severe eye

injury.

Impervious gloves of chemically resistant material (rubber or PVC) should be worn at all **Hand Protection** 

times. Wash contaminated clothing and dry thoroughly before reuse.

**Skin and Body Protection** Guidelines for hydrochloric acid, 37%

RECOMMENDED (resistance to breakthrough longer than 8 hours): Butyl rubber,

Neoprene rubber, Viton(TM), Viton(TM)/Butyl rubber, Barrier (PE/PA/PE), Trellchem(TM)

HPS, Trellchem(TM) VPS, Tychem(TM) SL (Saranex(TM)), Tychem(TM) CPF 3,

Tychem(TM) F, Tychem(TM) BR/LV, Tychem(TM) Responder(TM), Tychem(TM) TK. CAUTION, use for short periods only (resistance to breakthrough within 1 to 4 hours):

Polyethylene

NOT RECOMMENDED for use (resistance to breakthrough less than 1 hour): Polyvinyl

alcohol

**Respiratory Protection** 

NIOSH/OSHA RECOMMENDATIONS FOR HYDROGEN CHLORIDE (GAS)

CONCENTRATIONS IN AIR:

Up to 50 ppm: Chemical cartridge respirator with cartridge(s) to protect against hydrogeh chloride; or gas mask with canister to protect against hydrogen chloride; or powered airpurifying respirator with cartridge(s) to protect against hydrogen chloride; or powered airpurifying respirator with cartridge(s) to protect against hydrogen chloride; or SAR; or full-facepiece SCBA. Above this level, a full face self-contained breathing apparatus is required.

NIOSH approved acid gas or organic vapour cartridge(s) are required. EMERGENCY OR PLANNED ENTRY INTO UNKNOWN CONCENTRATION OR IDLH CONDITIONS: Positive pressure, full-facepiece SCBA; or positive pressure, full-facepiece SAR with an auxiliary positive pressure SCBA.

ESCAPE: Gas mask with acid gas canister; or escape-type SCBA.

**Thermal Hazards** 

Not Available

# Section 09 - Physical and Chemical Properties

#### <u>Appearance</u>

Physical State Fuming liquid

Colour Colourless or slightly yellow

Odour Pungent odour

Odour Threshold 1-5 ppm (detectable)

**Property** 

pH <1

Melting Point/Freezing Point -35°C

**Initial Boiling Point and Boiling** 

Range

62-90°C

Flash Point Not Applicable

Evaporation Rate <1

Flammability Non-flammable

Upper Flammable Limit Not Applicable

Lower Flammable Limit Not Applicable

Vapour Pressure (mm Hg, 20°C) 84 mmHg

Vapour Density (Air=1) 1.268 @ 20°C

Relative Density Not Available

Solubility(ies) Completely miscible

Partition Coefficient: n-

octanol/water

 $Log P_{ow} = 0.3$ 

Auto-ignition Temperature Not Applicable

**Decomposition Temperature** >1500°C

**Viscosity** 1.71-2.11 cSt

**Explosive Properties** In contact with metals, explosive hydrogen gas may form.

Specific Gravity (Water=1) 1.023-1.198

% Volatiles by Volume 100%

Formula HCI

Molecular Weight 34.46 g/mol

## Section 10 - Stability and Reactivity

Reactivity Contact with hypochlorites liberates chlorine gas. May react violently with incompatible

substances. Large amounts of heat can be released when concentrated hydrochloric acid is

mixed with water or with organic solvents.

**Stability** Stable, heat and contamination could cause decomposition.

**Possibility of Hazardous** 

Reactions

Hazardous polymerization does not occur.

**Conditions to Avoid** High temperatures. Incompatibles.

Incompatible Materials Metals, sodium, bases, formaldehyde, oxidizing agent, reducing agents, perchloric acid,

sulfuric acid. Potassium permanganate, aldehydes, epoxides, fluorine, acetylides, bories,

carbides, phosphide, silicides, hexalithium disilicide.

**Hazardous Decomposition** 

**Products** 

Contact with hypochlorites liberates chlorine gas. May react violently with incompatible substances. May release toxic and/or flammable gases such as hydrogen and phosphine

gas. Considerable amounts of heat may be evolved.

# Section 11 - Toxicological Information

#### **Acute Toxicity**

Component Oral  $LD_{50}$  Dermal  $LD_{50}$  Inhalation  $LC_{50}$ 

Hydrochloric Acid (35%) 2,121 mg/kg (rat) 4390 mg/kg (mouse) 1106 ppm (guinea pig, 4hr)

#### Chronic Toxicity – Carcinogenicity

Component IARC

Hydrochloric Acid Not classifiable as a human carcinogen.

**Skin Corrosion/Irritation**Corrosive. Contact may produce severe irritation or corrosive skin damage. Effects

range from dermatitis, photo sensitization, redness, swelling, pain, permanent scarring, to

death.

Causes severe burns of the mouth, esophagus, and stomach, with consequent pain, Ingestion

nausea, vomiting, diarrhea, circulatory collapse, and possibly death.

Inhalation Hydrochloric acid solutions can readily release high concentrations of hydrogen chloride

> gas, which is very toxic and corrosive and poses a serious inhalation hazard. Inhalation of even low concentrations is irritating and can cause coughing, pain, inflammation and swelling in the upper respiratory tract. A severe exposure can result in

a potentially fatal accumulation of fluid in the lungs (pulmonary edema). Symptoms of

pulmonary edema can be delayed for up to 24 or 48 hours after exposure.

Hydrochloric acid is corrosive to the eyes. Low concentrations of vapour or mist can be Serious Eye Damage/Irritation

irritating, causing redness. Concentrated vapour, mist or splashed liquid can cause

severe irritation and damage, burns and permanent blindness.

Hydrochloric acid is not considered an occupational respiratory or skin sensitizer. Respiratory or Skin Sensitization

The available evidence does not indicate that hydrochloric acid is a mutagen. **Germ Cell Mutagenicity** 

The limited evidence available does not indicate that hydrochloric acid is a developmental Reproductive Toxicity

toxin

Hydrochloric acid solutions release hydrogen chloride, a corrosive gas. Causes **STOT-Single Exposure** 

respiratory irritation.

Prolonged exposure can cause erosion and discolouration of teeth and chronic **STOT-Repeated Exposure** 

inflammation of nose, throat, and airways. In general, long-term skin contact with low concentrations of corrosive materials can cause dry, red, cracked skin (dermatitis).

Severe exposure can result in pulmonary edema and corrosion of tissues in the nose and **Aspiration Hazard** 

throat.

**Synergistic Materials** Not Available

## Section 12 – Ecological Information

**Ecotoxicity** 

Component **Toxicity to Algae Toxicity to Fish** Toxicity to Daphnia and Other Aquatic Invertebrates

Hydrochloric Acid EC<sub>50</sub>(Green algae, 72hr): LC<sub>50</sub>(Cyprinus carpio, 96 hr): 0.0492 mg/L

4.92 mg/L

LC<sub>50</sub>(Shrimp, 48hr): 100-300

ppm

Biodegradability Not Applicable - hydrochloric acid disassociates in water.

Hydrogen chloride does not accumulate in the food chain. Bioaccumulation

Mobility Hydrogen chloride dissociates into chloride and hydronium ions in moist soil.

Other Adverse Effects Extremely toxic to aquatic life by lowering the pH below 5.5. Dissociates in water and will

be neutralized by naturally occurring alkalinity and carbon dioxide. Acid will permeate soil,

dissolving soil material and will be neutralized somewhat.

# Section 13 – Disposal Considerations

Waste From Residues/Unused **Products** 

Dispose in accordance with all federal, provincial, and/or local regulations including the

Canadian Environmental Protection Act.

Dispose in accordance with all federal, provincial, and/or local regulations including the **Contaminated Packaging** 

Canadian Environmental Protection Act.

# Section 14 – Transport Information

**UN Number** UN1789

**UN Proper Shipping Name** HYDROCHLORIC ACID

8 Transport Hazard Class(es) Ш Packaging Group

Environmental Hazards Not listed as a marine pollutant under Canadian TDG Regulations, schedule III.

Special PrecautionsNot AvailableTransport in BulkNot Available

Additional Information Packing Group Limited Quantity Index

|| 1 L | | 1 L | | 5 L

TDG Other

Secure containers (full and/or empty) with suitable hold down devises during shipment and ensure all caps, valves, or closures are secured in the closed position.

TDG PRODUCT CLASSIFICATION: This product has been classified on the preparation date specified at section 14 of this MSDS / SDS, for transportation in accordance with the requirements of part 2 of the Transportation of Dangerous Goods Regulations. If applicable, testing and/or published test data regarding the classification of this product are listed in the references at section 16 of this MSDS / SDS.

# Section 15 - Regulatory Information

NOTE: THE PRODUCT LISTED ON THIS SDS HAS BEEN CLASSIFIED IN ACCORDANCE WITH THE HAZARD CRITERIA OF THE CANADIAN CONTROLLED PRODUCTS REGULATIONS. THIS SDS CONTAINS ALL INFORMATION REQUIRED BY THOSE REGULATIONS.

**NSF Certification**...... Product is Kosher certified. Product is certified under NSF/ANSI Standard 60 for pH adjustment and as a descaler at a maximum dosage for the following:

Hydrochloric Acid 10%: 140 mg/L	Hydrochloric Acid 19%: 74 mg/L	Hydrochloric Acid 28%: 50 mg/L
Hydrochloric Acid 11%: 127 mg/L	Hydrochloric Acid 20%: 70 mg/L	Hydrochloric Acid 29%: 48 mg/L
Hydrochloric Acid 12%: 117 mg/L	Hydrochloric Acid 21%: 67 mg/L	Hydrochloric Acid 30%: 47 mg/L
Hydrochloric Acid 13%: 108 mg/L	Hydrochloric Acid 22%: 64 mg/L	Hydrochloric Acid 31%: 45 mg/L
Hydrochloric Acid 14%: 100 mg/L	Hydrochloric Acid 23%: 61 mg/L	Hydrochloric Acid 32%: 44 mg/L
Hydrochloric Acid 15%: 93 mg/L	Hydrochloric Acid 24%: 58 mg/L	Hydrochloric Acid 33%: 42 mg/L
Hydrochloric Acid 16%: 88 mg/L	Hydrochloric Acid 25%: 56 mg/L	Hydrochloric Acid 34%: 41 mg/L
Hydrochloric Acid 17%: 82 mg/L	Hydrochloric Acid 26%: 54 mg/L	Hydrochloric Acid 35%: 40 mg/L
Hydrochloric Acid 18%: 78 mg/L	Hydrochloric Acid 27%: 52 mg/L	

NSF product use restrictions based on requirements obtained from the NSF website for current requirements.

#### Section 16 – Other Information

#### Preparation Date July 30, 2015

**Note:** The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations.

#### Attention: Receiver of the chemical goods / SDS coordinator

As part of our commitment to the Canadian Association of Chemical Distributors (CACD) Responsible Distribution<sup>®</sup> initiative, ClearTech Industries Inc. and its associated companies require, as a condition of sale, that you forward the attached Safety Data Sheet(s) to all affected employees, customers, and end-users. ClearTech will send any available supplementary handling, health, and safety information to you at your request.

If you have any questions or concerns please call our customer service center.

#### References:

1) CHEMINFO

- 2) eChemPortal
- 3) TOXNET
- 4) Transportation of Dangerous Goods Canada
- 5) HSDB
- 6) ECHA

# **ClearTech Industries Inc. - Locations**

Corporate Head Office: 1500 Quebec Avenue, Saskatoon, SK, S7K 1V7
Phone: 1(306) 664 – 2522
Fax: 1(888) 281-8109

www.cleartech.ca

24 Hour Emergency Number - All Locations - 1(306) 664-2522